The Infra-red Spectrum and Vibrational Assignments of Poly(phosphonitrilic chloride)

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The infra-red spectrum of poly(phosphonitrilic chloride) has been obtained in the region 5000-20 cm⁻¹. Correlation of the results with the spectra of some phosphonitrilic halide oligomers has enabled characterization of the polymer spectrum to be made and assignments for the phosphorus-halogen and skeletal vibrational modes are given. As expected most of the fundamentals of this molecule appear to lie in the far infra-red region. No line group splitting of these bands was observed. The vibrational spectrum of the polymer supports a distorted helical structure of C₂ symmetry.

CYCLIC phosphonitrilic chloride and phosphonitrilic bromide oligomers may be readily converted into a high polymer of the form $(NPX_2)_n$, where X=Cl or Br. Little work appears in the literature on physical methods of analysis of the inorganic polymer $(NPCl_2)_n$ and we are not aware of any infra-red spectroscopic studies. The X-ray pattern of the fibre which crystallizes on stretching was first investigated by Meyer *et al.*¹, who proposed a linear backbone chain of regular conformation with two mono-





meric units in the identity period, and deduced an orthorhombic unit cell. Huggins² later proposed a uniform helical conformation with two monomeric units in the repeat unit [*Figure 1* (a)], using the electron diffraction results of Brockway and Bright³ for the molecular structure of $(NPCl_2)_3$.

More recently, Giglio et al.⁴ have shown that the identity period is not sufficient to define the chain conformation, as the study of many organic high polymers has shown. They interpreted the X-ray fibre results of stretched (NPCl₂), by means of Fourier transform methods in an attempt to establish the chain conformation without considering the mode of packing in the crystal lattice. They concluded that the analysis was nearly consistent with a twofold helical conformation which they designate 'cisplan' [Figure 1 (b)], the values for the angles of rotation around the P-N and N-P bonds being 156° and 14° respectively. The 'cis-plan' conformation would correspond to angles of rotation of 180° and 0°, respectively. The analysis yielded a P-N bond length for the linear polymer of 1.60 Å, and bond angles $\hat{PNP} = 127^{\circ}$. $\hat{NPN} = 119^{\circ}$ and P-Cl = 1.99 Å, $ClPCl = 102^{\circ}$. These workers also concluded that the bond angles of the backbone chain in the linear polymer, although not accurately determined in their analysis, were slightly different from those established for (NPCl₂)₃ in which $\hat{NPN} = \hat{PNP} = 120^\circ$.

In this investigation the infra-red spectrum of poly(phosphonitrilic chloride) has been examined for both the stretched polymer, and for polymer cast as a thin film on a transparent substrate, and for rubber swollen in carbon tetrachloride (CCl₄) and chloroform (CHCl₃). A comparison of the polymer spectrum with the spectra of the oligomers (NPCl₂)₃^{5,6}, (NPBr₂)₃^{7,8}, (NPCl₂)₄^{7,8} and (NPBr₂)₄^{7,8} has enabled some characterization of the polymer to be made.

EXPERIMENTAL

Preparation of pure $(NPCl_2)_n$

The polymer was initially prepared as described by Meyer *et al.*¹. Commercial $(NPCl_2)_3$ (Albright and Wilson) which had been recrystallized from CCl₄ and then sublimed, was heated at 300°C for twelve hours in a glass capsule sealed under vacuum. A brownish-black rubbery plug which could only be removed from the capsule with difficulty was obtained in good yield. After repeating the preparation several times with further purified oligomer over a range of temperatures between 250° and 325°C, the product continued to be a dark rubber.

It was concluded that the 'purified' oligomer used in the preparation still contained traces of an impurity which was characterized by giving a brownish-black polymer, and by being a good catalyst at about 250°C. Subsequently, a sample of poly(phosphonitrilic chloride) was obtained by courtesy of Dr R. O. Colclough. This was very pure and easily worked and was used in some runs. It appeared to be of lower molecular weight and less branched than our material but no significant difference in the spectra was noticed. We concluded that any impurities in our own polymer were not affecting the spectrum. The low molecular weight polymer was prepared by recrystallizing the oligomeric (NPCl₂), from CCl₂, followed by refluxing over barium oxide (BaO), and then subliming the product several times, the penultimate sublimation being carried out over a phosphorus pentoxide (P_2O_5) boat, to remove all traces of moisture. The latter can also catalyse polymerization at 250°C. On heating such a sample of (NPCl₂)₃ in vacuo at 260°C for 72 hours, the product was a milky-white rubbery plug which, when swollen in CCl₄ to remove unconverted oligomer, vielded a colourless polymer⁹

Sampling techniques

In the far infra-red region, three sampling techniques were employed. The 'dry' rubber, obtained by removing all traces of CCl, accumulated during swelling, and removal of unconverted oligomer, was stretched and clamped between high density (Rigidex) polyethylene plates. It was found possible to 'cast' thin films of low molecular weight rubber on to a polyethylene substrate by 'smearing' the latter with rubber swollen in CCl.. The solvent was removed under vacuum and the film clamped (and stretched) between two polyethylene discs, to prevent peeling. High molecular weight polymer was also swollen in CCl4 and CHCl3, and sections were microtomed for examination between polyethylene windows. The



Figure 2—Infra-red spectrum of poly(phosphonitrilic chloride) in the region $2000-500 \text{ cm}^{-1}$

three types of sample were examined on a Grubb Parsons-NPL interferometer and the Grubb Parsons DM4 and GM3 far infra-red grating spectrometers. The far infra-red spectrum of $(NPCl_2)_n$ film, cast and then stretched on to polyethylene is shown in Figure 3. The conventional infrared spectrum is shown in Figure 2.

In the conventional region much thinner specimens of polymer had to be used to produce clear spectra. The three sampling procedures described above were also adapted in this region, substituting potassium bromide windows for the polyethylene plates. The physical properties of the linear



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 $(NPC1_{2})_n$ rubber made it almost impossible to obtain oriented specimens of suitable thickness by mechanical treatment. The polymer loses its elastic properties on standing at room temperature (thin films are also particularly vulnerable to hydrolysis), the process being accelerated whilst in the beam of the spectrometer and for this reason a study with polarized radiation using oriented samples could not be made.

Assignment of normal modes of vibration of $(NPC1_2)_n$

In attempting to make an assignment of the normal vibrations of the polymer, the two most probable structures for the isolated polymer chain will be considered. These are the normal helical model proposed by $\hat{P}_{N} = 1.00^{\circ}$, and $\hat{P}_{N} = 1.00^{\circ}$, and the distorted 'cis-plan' helical model of Giglio *et al.*⁴ [Figure 1 (b), with P-N=1.60 Å, $\hat{P}NP=127^{\circ}$ and $\hat{N}PN=119^{\circ}$]. The infra-red spectral results will then be applied to the predicted spectrum of each structure in turn in order to assess the best 'structural fit'.

Normal helical model (Figure 1 (a))

The normal helical model contains two monomeric units in the polymer identity period which constitutes one turn of the helix⁴. The hypothetically infinite and isolated polymer chain possesses four symmetry elements, namely the identity, a twofold screw axis about the fibre axis, and two

twofold axes bisecting the PNP and NPN angles, and perpendicular to the screw axis. The factor group of this line group will then be isomorphous to the point group D_2 . Since the polymer identity period

$$[-N=(PCl_2)-N=(PCl_2)-]$$

contains eight atoms, the fragment will possess (3N-4) or 20 normal modes of vibration. According to Table 1, in the isolated chain, these

Species	E	$C^{1}(z)^{*}$	$C_2(y)$	$C_2(x)$	N^{\dagger}			
	1	1	1	1	5		$\alpha_{xx}\alpha_{yy}\alpha_{zz}$	
B ₁	1	1	-1	-1	5	$T_{z}(), R_{z}$	α_{xy}	
B_2	1	-1	1	-1	5	$T_{y}(\perp), R_{y}$	α_{zx}	
B_{3}	1	1	-1	1	5	$\tilde{T}_{x}(\perp), \tilde{R}_{x}$	α_{yz}	

Table 1. Characters, numbers of normal modes and selection rules for poly(phosphonitrilic chloride) under the group D_2

The symmetry operation $C^{1}(z)$ is a rotation of angle π about the axis of the helix followed by a translation along the fibre axis of half the unit cell length.

taking along the note axis of that the unit cert fengul. $\uparrow N$ is the number of total normal modes under each symmetry species. fT (I) and $T(\perp)$ are pure translations (denoting infra-red activity) parallel and perpendicular to the helical axis.

R denotes pure rotation about the helical axis.

modes will be distributed as follows:

$$5A + 5B_1 + 5B_2 + 5B_3$$

where the A modes are Raman active, and the B_1 , B_2 and B_3 modes are both infra-red and Raman active. The B_1 modes give rise to a transition

moment parallel to the helical axis (z axis) and the B_2 and B_3 modes give rise to transition moments perpendicular to the helical axis. In the crystal the only symmetry operations which leave the axis of the chain invariant are the factor group operations E and $C_2(z)$. Consequently the site group in the crystal will be isomorphous to the C_2 point group.

Distorted 'cis-plan' helical model, Figure 1 (b)

The distorted 'cis-plan' helical structure will also contain two monomeric units in the polymer identity period⁴ and will comprise one turn of the helix. The isolated polymer considered as an infinite chain possesses only two symmetry elements, namely the identity and the twofold screw axis about the fibre axis (z axis). The factor group will then be isomorphous to the point group C_2 (also the line group C_2) and according to the characters in *Table 2*, the twenty normal modes of vibration in the isolated chain will then divide as follows

10A + 10B

where both the A and B modes are infra-red and Raman active. The site group in the crystal will be isomorphous to the C_2 point group, as in the normal helical model.

Table 2. Characters, numbers of normal modes and selection rules for $(NPCl_2)_n$, under the group C_2

Species	E	$C^{1}(z)$	N	·	····
A	1	1	10	$\overline{T_z, R_z}$	$\alpha_{xx} \alpha_{yy} \alpha_{zz} \alpha_{xy}$
В	1	-1	10	T_x, T_y	$\alpha_{yz}\alpha_{zx}$
				R_x, R_y	

Discussion of the vibrational spectrum

Of the 30 bands definitely resolved in the spectrum of the stretched linear polymer on transparent substrates there are 23 in the region where fundamental modes are likely to occur (*Table 3*).

Since the distorted 'cis-plan' model predicts 20 infra-red-active fundamentals and the normal helical model only 15, the infra-red spectral evidence favours the former structure. Further support for the distorted helical model lies in the fact that four bands which could be readily attributed to stretching of the P—N bond are observed. The distorted helix predicts four infra-red-active P—N stretching modes whilst the normal helix predicts only three.

A comparison of the crystal spectra of $(NPCl_2)_3$, $(NPCl_2)_4$ and the 'cast film' spectrum of $(NPCl_2)_n$ reveals that there are a number of bands lying in the far infra-red spectrum common to all three molecules and it is felt that band correlations can be made in most of these cases between the oligomeric and the polymeric vibrational modes. The common bands are nearly all believed to be due essentially to PCl₂ motion. Using these correlations, a tentative assignment for the PCl₂ vibrational modes has been made for the linear polymer, and these are summarized in *Table 4*.

Rubber swollen in CCl ₄	Film stretched on substrate	Assignment*
	49 vvw	
	62 vvw	
80 vvw (77 vvw \	$\tau = v(A)$
91 vvw [}]	89 vvw [∫]	skeleton ^v 10 ⁽²¹⁾
	105 vvw	
	128 vvw	$\tau \mathrm{PCl}_{2}, \nu_{20}(B)$
155 vvw	157 vw	$\rho \mathrm{PCl}_2 + \delta \mathrm{PNP}, \nu_{15}(B)$
168 vvw	172 vw	$\tau \mathrm{PCl}_2, \nu_5(A)$
183 vvw	185 vw	$\rho \mathrm{PCl}_2 + \nu \mathrm{PN}, \nu_9(A)$
	222 vvw	
268 w	262 w	$\delta PCl_2, \nu_{19}(B)$
	336 sh	$\delta PCl_2 / \omega PCl_2$
	368 sh	$\nu_{\rm s} {\rm PCl}_2 / \omega {\rm PCl}_2$
400 m. b	397 m	$\nu_{\rm s} {\rm PCl}_2, \nu_{18}(B)$
460 m, b	464 m	$\delta PNP, \nu_{13}(B)$
513 w, b	515 w, b	$\nu_{\rm as} {\rm PCl}_2, \nu_7(A)$
580 s, b	565 s, b	$\nu_{\rm as} {\rm PCl}_2, \nu_{12}(B)$
750 m	745 m	δ NPN, $\nu_{17}(B)$
821 vvw	815 vvw	$565 (\nu_{12}, B) + 262 (\nu_{19}, B) = 827$
863 vw	872 vw	$\delta \text{NPN}, \nu_2(A)$
	1 1 94 sh	$\nu \mathbf{P} = \mathbf{N} + \rho \mathbf{PCl}_2, \nu_{11}(B)$
1 225 vs, b	1 230 vs	$\nu \mathbf{P} = \mathbf{N} + \rho \mathbf{PCl}_2, \nu_6(A)$
	1 275 vs	$\nu_{\rm s} \mathbf{P} = \mathbf{N}, \nu_{\rm l}(A)$
1 304 vs, b	1 300 vs, b	$\nu \mathbf{P} = \mathbf{N} + \rho \mathbf{PCl}_2, \nu_{16}(\mathbf{B})$
1 701 sh	1 700 sh	$1\ 300\ (\nu_{16},\ B)+397\ (\nu_{18},\ B)=1\ 697$
1 768 w†	1 755 vw, b†	$1\ 300\ (\nu_{18}, B) + 464\ (\nu_{13}, B) = 1\ 764$
1 981 sh	1 985 sh	$1\ 230\ (\nu_6, A) + 745\ (\nu_{17}, B) = 1\ 975$
2 120 m†	2 120 vvw, b†	$1\ 275\ (\nu_1, A)\ +464\ (\nu_{13}, B)\ +397\ (\nu_{18}B)\ =\ 2\ 136$
2 930 vw	2 910 vw	

Table 3. Infra-red spectra of poly(phosphonitrilic chloride)

*The assignments are based on the distorted 'cis-plan' helical structure for $(NPC1_{2n})$, *Bands much stronger in highly branched rubber.

Also included in *Table 4* are approximate descriptions of the normal vibrations involved and the form of their division under the A, B_1 , B_2 and B_3 species of the D_2 (normal helix) and the A and B species of the C_2 (distorted helix) line groups. It should be noted that the description of the vibrational modes in terms of simplified group frequencies is very approximate, since appreciable mixing of the various symmetry coordinates does occur for simpler molecules, especially for vibrations occurring at frequencies below 1 300 cm⁻¹.

Assignment of PCl₂ modes

(i) PCl₂ stretching modes—Four PCl₂ stretching modes are expected, namely $\nu_s PCl_2$ (in-phase), $\nu_s PCl_2$ (out-of-phase), $\nu_{as} PCl_2$ (in-phase) and $\nu_{as} PCl_2$ (out-of-phase). In (NPCl₂)₃, $\nu_{as} PCl_2$ (in-phase) occurs strongly at 544 cm⁻¹, and in (NPCl₂)₄ it occurs at 521 cm⁻¹. Consequently the weak, broad absorption at 515 cm⁻¹ is assigned to this mode (ν_7) in (NPCl₂)_n.

				emeride)	
Line group species (D ₂)	Line group species (C ₂)	No.	P	ν _i	Approximate motion
A (R)	A (R, IR)	1		1 275	Sym. P=N stretching
		2		872	NPN skeletal deformation
		3		397/368	* PCl, sym. stretch, in-phase
		4		336*	PCl ₂ deformation, in-phase
		5		172	PCl_2 twisting, out-of-phase
$\overline{B_1(\mathbf{R},\mathbf{IR})}$	A (R, IR)	6	π	1 230	P=N sym. stretching+PCl ₂
		7	π	515	PCl_2 asym. stretch, in-phase
		8	π	.336*	PCl, wagging, out-of-phase
		9	π	185	PCl ₂ rocking+skeletal stretching
		10	π	89	skeletal torsion
$B_2(\mathbf{R}, \mathbf{IR})$	$B(\mathbf{R},\mathbf{IR})$	11	σ	1 194	P-N stretching + PCl ₂ rocking
		12	σ	565	PCl ₂ asym. stretching, out-of-phase
		13	σ	464	PNP skeletal deformation
		14	σ	368/336	*PCl ₂ wagging, out-of-phase
		15	σ	157	PCl ₂ rocking + PNP deformation
$\overline{B_3(\mathbf{R},\mathbf{IR})}$	<i>B</i> (R, IR)	16	σ	1 300	P=N stretching+PCl ₂ rocking
		17	σ	745	NPN skeletal deformation
		18	σ	397*	PCl ₂ sym. stretching, out-of-phase
		19	σ	262	PCl ₂ deformation, out-of-phase
		20	σ	128	PCl ₂ twisting, in-phase

Table 4. Assignment of the fundamental modes of vibration of poly(phosphonitrilic chloride)

*Several alternative assignments are possible for these bands; see text.

Similarly, the broad, strong absorption in $(NPCl_2)_n$ at 565 cm⁻¹ (cast film) and ~ 580 cm⁻¹ (rubber swollen in CCl₄ is assigned to v_{as} PCl₂ (out-ofphase), ν_{12} . In (NPCl₂)₃ the mode occurs strongly at 582 cm⁻¹ and in $(NPCl_2)_4$ at 578 cm⁻¹ The symmetric out-of-phase stretching mode, $\nu_s PCl_2$ occurs as a medium band at 404 cm⁻¹ in $(NPCl_2)_3$ and at 390 cm⁻¹ in (NPCl₂)_a and the medium band at 397 cm⁻¹ in (NPCl₂)_n is therefore reasonably attributed to this vibration (ν_{18}). The totally symmetric $\nu_s PCl_2$ (inphase) mode in $(NPC1_2)_n$ is less readily assigned. In $(NPC1_2)_3$ the mode is believed to occur at 365 cm⁻¹ (Raman) and in (NPCl₂), at 404 cm⁻¹ (Raman). The vibration is expected to occur only weakly in the infrared in (NPCl₂)_n for the distorted C_2 structure and could easily be masked by the medium absorption at 394 cm⁻¹. A pronounced shoulder occurs at 368 cm⁻¹ which could also be attributed to this mode. Since the extent of mixing of this vibration with the chain skeleton is unknown in the linear polymer, no assignment can be made with confidence by direct correlation. This mode will be inactive in the normal helical model of (NPCl₂),

(ii) PCl_2 wagging modes—The two PCl_3 'wagging' modes are the most difficult PCl_2 vibrations to locate, since appreciable mixing of the symmetry coordinates is likely to occur. In $(NPCl_2)_3$ the ωPCl_2 modes are believed

to occur in the region 320 to 340 cm⁻¹ and in (NPCl₂)₄ in the region 340 to 420 cm⁻¹. The only band which can be reasonably assigned to an ω PCl₂ mode in (NPCl₂)_n is the shoulder at 336 cm⁻¹. The medium band at 397 cm⁻¹ is very broad on the longer-wave side and it is felt that other bands, as yet unresolved, may contribute to this effect. It seems probable that the other PCl₂ wagging vibration also contributes to this broad shoulder.

(iii) PCl₂ deformation modes—The out-of-phase δ PCl₂ mode (ν_{19}) is assigned without difficulty in (NPCl₂)_n to the weak band at 262 cm⁻¹. It occurs characteristically in (NPCl₂)₃ and (NPCl₂)₄ at 260 cm⁻¹, the extent of mixing being only slight. The totally symmetric in-phase δ PCl₂ modes in (NPCl₂)₃ and (NPCl₂)₄ occur as polarized Raman bands at 304 and 352 cm⁻¹ respectively. In (NPCl₂)_n it is felt that the broad unresolved shoulder in the region 380 to 300 cm⁻¹ contributes to this mode (ν_4).

(iv) PCl_2 torsional modes—Characteristic absorption at 177 cm⁻¹ in $(NPCl_2)_3$ and 170 cm⁻¹ in $(NPCl_2)_4$ has been attributed to out-of-phase PCl_2 torsion, τPCl_2 . Since the forms of normal vibration and the extent of coordinate mixing in these τPCl_2 modes appears to be similar in the oligomers and the polymer, the weak absorption at 172 cm⁻¹ in $(NPCl_2)_n$ is also assigned to this mode in $(NPCl_2)_n$ (ν_5). This vibration is only infrared active in the distorted helical model for the linear polymer, so if it is correct then it is strong evidence for the presence of the distorted C_2 structure.

The in-phase PCl₂ torsions in $(NPCl_2)_3$ and $(NPCl_2)_4$ are believed to occur at 131 and 137 cm⁻¹, respectively. An extremely weak band at 128 cm⁻¹ was observed in $(NPCl_2)_n$ and may be attributable to ν_{20} , τPCl_2 (in-phase).

PCl₂ rocking modes—The mixed vibrations due to out-of-phase $[\rho PCl_2 + \delta PNP]$ in $(NPCl_2)_3$ and $(NPCl_2)_4$ are believed to occur at 162 and 158 cm⁻¹ respectively. By analogy, the mode in $(NPCl_2)_n$ (ν_{15}) is assigned to the very weak absorption at 157 cm⁻¹.

The assignment of the 'mixed' in-phase PCl₂ rocking mode (ν_9) is in some doubt. A band common to both (NPCl₂), and (NPCl₂)_n occurs quite distinctly at 88 cm⁻¹, but is not found in (NPCl₂)₃. Since it is the lowest observed fundamental in (NPCl₂)₄ it could reasonably be attributed to ρ PCl₂ (in-phase). However, in (NPCl₂)₃ the lowest fundamental is found at 131 cm⁻¹ and this has been assigned to τ PCl₂ (in-phase), although it is possible that a lower fundamental remains undetected in this molecule. This PCl₂ rocking mode, which occurs as a mixed (ρ PCl₂ + skeletal stretching) mode is tentatively assigned to the very weak band at 185 cm⁻¹, where the mode is believed to occur in (NPCl₂)₈.

Skeletal modes

(i) P=N stretching modes—The intense, broad absorption in (NPCl₂)_n at $\sim 1300 \text{ cm}^{-1}$ is resolved in very thin film specimens into four components at 1 300, 1 275, 1 230 and 1 194 cm⁻¹. There are four infra-red active P=N stretching modes expected in the distorted helical model,

only one of these involving essentially symmetric stretching of the P=N bond. The other three vibrations are 'mixed' modes involving

$$(\nu P = N + \rho P C l_2), \nu_6, \nu_{11} \text{ and } \nu_{16}.$$

From a knowledge of the mixed skeletal stretching vibrations occurring in other similar systems¹⁰ we would tentatively assign these frequencies in the order ν_{16} (1 300) $> \nu_1$ (1 275) $> \nu_6$ (1 230) $> \nu_{11}$ (1 195). These $\nu P = N$ frequencies are somewhat lower than those observed in (NPCl₂)₄ (which occur in the region 1 360 to 1 300 cm⁻¹) and (NPCl₂)₃ (1 370, 1 226 cm⁻¹) which is due to the mixing with the PCl₂ rocking modes, which would appear to occur to a greater extent in the linear polymer.

(ii) PNP and NPN skeletal deformation modes—A comparison of the infra-red and Raman spectra of $(NPCl_2)_3$ and $(NPCl_2)_4$ with the infra-red spectrum of $(NPCl_2)_n$ reveals that the strong bands assigned to ring ('skeletal') breathing, in-plane deformation and elongation vibrations in the trimer at 877, 785, 671, 601 and 533 cm⁻¹ and in the tetramer 894, 794, 752, 725, 667, 602, 538 and 434 cm⁻¹, are absent from the spectrum of the linear polymer. However, it is thought that the infra-red bands occurring in the latter at 872, 815, 745 and 464 cm⁻¹ must be due to skeletal modes since they certainly cannot be accounted for as vibrations involving primarily PCl₂ motion.

Due to the lack of polarization measurements, and since a normal coordinate analysis has not been performed on this molecule, there is insufficient evidence for an unambiguous assignment of these skeletal deformation modes. Consequently, their assignment as given in Table 4 is somewhat arbitrary. However, in orthorhombic polyoxymethylene, which is believed to have a similar helical structure to $(NPCl_2)_n$, characteristic OCO and COC deformation modes (equivalent to v_{17} and v_{18} in Table 4) do appear strongly in the infra-red at 594 and 430 cm⁻¹ respectively¹⁰. By analogy the infra-red bands of medium intensity at 745 and 464 cm⁻¹ in (NPCl₂)_n (all the remaining unassigned bands are very weak) are tentatively assigned to $\delta NPN(\nu_{17})$ and $\delta PNP(\nu_{13})$, respectively. The band at 872 cm⁻¹ is tentatively attributed to $\delta NPN(A)$, ν_2 . If the latter assignment were correct then the appearance of this band would be further evidence of the distorted C_2 helical structure for (NPCl₂)_n. The remaining band unassigned in the region 1000 to 400 cm⁻¹, namely the extremely weak absorption at 815 cm⁻¹, could be accounted for as the combination tone

$$565 + 262 = 827$$
 cm⁻¹

(iii) Skeletal torsional mode—The torsional vibration of the —P—N—P—N— chain is expected at a very low frequency. Bands in the far infra-red at 89, 77, 62 and 49 cm⁻¹ are observed in the spectrum of the stretched film of $(NPCl_2)_n$. Of these the most intense is the absorption at 89 cm⁻¹ and this band is therefore tentatively assigned to the skeletal torsional mode, ν_{10} . However, it is possible that the absorption complex at 77, 62 and 49 cm⁻¹ is in some way associated with the torsional oscillation of the chain backbone.

Assignment of combination bands

The weak bands as yet unaccounted for at 1 700, 1 755, 1 985 and 2 120 cm⁻¹ can only reasonably be accounted for in terms of the following combination tones:

 $\begin{array}{c} 1300 \ (\nu_{16}) + 397 \ (\nu_{18}) = 1 \ 697 \ (\text{Observed } 1 \ 700) \\ 1300 \ (\nu_{16}) + 464 \ (\nu_{13}) = 1 \ 764 \ (\text{Observed } 1 \ 755) \\ 1230 \ (\nu_6) + 745 \ (\nu_{17}) = 1 \ 975 \ (\text{Observed } 1 \ 985) \\ 1275 \ (\nu_1) + 464 \ (\nu_{13}) + 397 \ (\nu_{18}) = 2 \ 136 \ (\text{Observed } 2 \ 120) \end{array}$

Discussion

No X-ray information on the number of polymer chains expected in the unit cell in the polymer crystal is available. However, theory predicts that if N polymer chains are placed in the crystal lattice, each of the line group modes should split into N components. In the distorted helical structure all the components arising due to A and B line group splitting would be infra-red active, provided that the intermolecular coupling constants in the crystal are large enough to produce appreciable separation of the energy levels. In the normal helical structure, if N is even, only N/2 of the components arising from the A and B_1 line group modes, and all N components of the B_2 and B_3 line group modes will be infra-red active.

No such splitting of any of the bands in the stretched film spectrum of the polymer was observed. There was only slight improvement in resolution in the stretched film spectrum compared to the spectrum of the polymer swollen in CCl₄. No definite conclusions may be drawn from this observation at this stage since the physical nature of the polymer is such that, although stretching does induce crystallization⁴ the extent of crystallization in the stretched rubber diminishes over a period of time and is temperature dependent. In addition, even if N>1 the intermolecular coupling in such a sample at room temperature might not be great enough to cause splitting of the vibrational levels.

Although the preliminary evidence present in this study favours the distorted 'cis-plan' helical structure, definite conclusions await the examination of the stretched film in polarized infra-red radiation and at liquid nitrogen temperatures. It is hoped that the experimental difficulties associated with such a study will shortly be overcome, and it is also intended to perform a normal coordinate analysis on the polymer.

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ANNOUNCEMENT

SYMPOSIUM ON 'POLYMERS AND WATER'

The annual Polymer Symposium organized by the Department of Chemistry of the University of Wales Institute of Science and Technology will be held on 4 July and will be concerned with water-soluble polymers and the properties of polymer-water systems. The symposium will be of interest to those concerned with adhesives, packaging, textiles and food science. The following subjects will be discussed under the leadership of the named invited speakers: Polyvinyl alcohol—Dr C. A. FINCH; Gelatin and the gelatin-water system—Dr G. STAINSBY; Sorption and diffusion of water polymers—Dr J. A. BARRIE; and The electrical properties of polymerwater systems—Dr J. A. MEDLEY. Further details are obtainable from the Organizer of Short Courses, U.W.I.S.T., 22 North Road, Cardiff, CF1 3DY.